# Oxidation of Hydrogen and Reduction of Methanol to Methane is the Sole Energy Source for a Methanogen Isolated from Human Feces

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A methanogenic coccus isolated from human feces requires  $H_2$  and  $CH_3OH$  for growth and uses  $H_2$  to reduce  $CH_3OH$  to  $CH_4$ . Growth does not occur with  $CH_3OH$  alone. The organism does not grow or produce  $CH_4$  from acetate or methylamines without or with  $H_2$  or from  $H_2$  and  $CO_2$  or formate. In a complex medium,  $CO_2$  is required for formation of approximately 50% of cell carbon, whereas the methyl carbon from methanol is not incorporated into cell carbon.

Almost all methanogens obtain energy for growth by using  $H_2$  to reduce  $CO_2$  to  $CH_4$  (1). Some methanogens obtain energy by producing  $CH_4$  from the methyl group of acetate,  $CH_3OH$ , or methylamines (1). The family *Methanosarcinaceae* contains the best known examples of methanogens that produce  $CH_4$  from methyl groups. Conversion of  $CH_3OH$  to  $CH_4$  proceeds according to the following equation:

$$4CH_3OH \rightarrow 3CH_4 + CO_2 + 2H_2O$$
 (1)

The electrons removed during the oxidation of 1 mol of  $CH_3OH$  to  $CO_2$  are used to reduce 3 mol of  $CH_3OH$  to methane and  $H_2O$ .

We recently isolated a methanogenic coccus from an anaerobic enrichment of human feces in a medium that contained  $CH_3OH$  as the substrate for  $CH_4$  production. The energy metabolism of the isolate was unique when compared to those of all previously studied methanogens. The results presented in this report show that the organism grew only when it was provided with both  $H_2$  and  $CH_3OH$ . It used  $H_2$  to reduce  $CH_3OH$  to  $CH_4$  and  $H_2O$  according to the following equation:

$$H_2 + CH_3OH \rightarrow CH_4 + H_2O \qquad (2)$$

The isolate did not grow on acetate or methylamines, with or without  $H_2$ , or by using  $H_2$  to reduce  $CO_2$  to  $CH_4$ .

## MATERIALS AND METHODS

Organism. An anaerobic enrichment culture that produced CH<sub>4</sub> from CH<sub>3</sub>OH had been established with an inoculum of human feces and was transferred monthly. Microscopic examination showed a large coccus, which had factor 420 fluorescence when visualized with epifluorescence microscopy, as described previously (6). Repeated attempts to isolate the coccus

were unsuccessful with CH<sub>3</sub>OH and 80%  $N_2$ –20% CO<sub>2</sub> ( $N_2$ -CO<sub>2</sub>) as the gas phase. However, when the enrichment was plated on a medium which contained 0.4% CH<sub>3</sub>OH and 101.3 kPa of 80% H<sub>2</sub>–20% CO<sub>2</sub> (H<sub>2</sub>-CO<sub>2</sub>), a pure culture of a fluorescent CH<sub>4</sub>-producing coccus was isolated that was morphologically similar to the fluorescent cocci in the enrichment culture. The isolate was usually in pairs or tetrads and what appeared to be clumped packets of cocci (Fig. 1). Details of the enrichment procedure and microbiological characterization of the isolate will be published separately.

Growth studies. The serum bottle modification of the Hungate technique for cultivating anaerobic bacteria was used to prepare, dispense, and autoclave media and to transfer cultures (4). All incubations were at 37°C, and liquid cultures were agitated by shaking or

The isolate was tested for its ability to grow and produce CH<sub>4</sub> from H<sub>2</sub>-CO<sub>2</sub>, formate, acetate, CH<sub>3</sub>OH, and trimethylamine as described previously (6). Growth and CH<sub>4</sub> production with 0.5% methylamine, 0.5% dimethylamine, and 0.76% ethanol were examined under the same conditions.

The effect of CO<sub>2</sub> and H<sub>2</sub> on growth with CH<sub>3</sub>OH and the stoichoimetry of CH<sub>4</sub> production were examined with cultures grown in a phosphate-buffered medium. It contained (per liter): Trypticase (BBL Microbiology Systems, Cockeysville, Md.) and yeast extract (Difco Laboratories, Detroit, Mich.), 2.0 g each; sodium acetate and sodium formate, 0.5 g each; thiamine-hydrochloride, nicotinamide, riboflavin, pyridoxine-hydrochloride, and calcium D-pantothenate. 2.0 mg each; cyanocobalamin, 0.2 mg; biotin, 10 mg; p-aminobenzoic acid, 1.0 mg; folic acid, 0.5 mg; Fe- $SO_4 \cdot 7H_2O$ , 1.2 mg;  $K_2HPO_4$ , 0.3 g;  $KH_2PO_4$ , 2.8 g;  $\begin{array}{l} (NH_4)_2SO_4,\,0.3\;g;\,NaCl,\,0.61\;g;\,MgSO_4\cdot7H_2O,\,0.153\\ g;\,CaCl_2\cdot2H_2O,\,76\;mg;\,MnSO_4\cdot7H_2O,\,5\;mg;\,CoSO_4 \end{array}$ and ZnSO<sub>4</sub>, 1 mg each; CuSO<sub>4</sub>  $\cdot$  5H<sub>2</sub>O, AlK(SO<sub>4</sub>)<sub>2</sub>. H<sub>3</sub>BO<sub>3</sub>, and Na<sub>2</sub>MoO<sub>4</sub>  $\cdot$  2H<sub>2</sub>O, 100  $\mu$ g each; nitrilotriacetic acid, 15 mg; NH<sub>4</sub>Cl, 1.0 g; L-cysteine hydrochloride, 0.875 g; Na<sub>2</sub>S · 10 H<sub>2</sub>O, 0.375 g; and resazurin, 1 mg. The pH was adjusted to 7.0 before boiling. After rapid cooling, the medium was gassed with 100%

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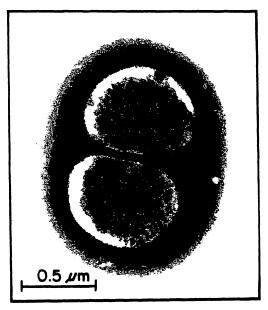




FIG. 1. Electron micrographs of cocci from a pure culture grown on 0.3% CH<sub>3</sub>OH and 202.6 kPa of H<sub>2</sub>-CO<sub>2</sub> in phosphate-buffered medium. The OD at 660 nm of the 48-h culture was 1.3. Cells on carbon- and Formvar-coated copper grids were negatively stained with a 2% aqueous solution of ammonium molybdate.

 $N_2$ . Cysteine and sulfide were added as previously described (4).

All turbidity (optical density [OD]) measurements were made with 5-ml cultures in 18- by 150-mm serum-finished tubes (Bellco Glass, Inc., Vineland, N.J.).

Measurements against an uninoculated medium blank were made in a Bausch & Lomb Spectronic 70 spectrophotometer at 660 nm.

14C labeling experiments. The headspaces of serum bottles of known total volume containing reduces

TABLE 1. Dependence of growth of isolate on H<sub>2</sub> and CH<sub>3</sub>OH

Gas phase	OD at 660 nm with"		
	No addition	CH <sub>3</sub> OH"	
H <sub>2</sub> -CO <sub>2</sub> N <sub>2</sub> -CO <sub>2</sub>	0.27 0.10	1.90 0.17	

<sup>&</sup>quot;Highest OD during 8 days of incubation. These occurred after 96 h for the H<sub>2</sub>-CO<sub>2</sub> plus CH<sub>3</sub>OH culture and after 24 h for the remaining cultures.

phosphate-buffered medium were gassed with  $H_2$ - $CO_2$  for 10 min. Then  $^{14}CH_3OH$  (1.86  $\mu$ Ci; ca. 5,000  $\mu$ mol) was added aseptically to one bottle, and cold  $CH_3OH$  (ca. 5,000  $\mu$ mol) and  $[^{14}C]$ sodium bicarbonate (1.97  $\mu$ Ci; 0.73  $\mu$ mol) were aseptically added to the other.

Each bottle was inoculated with 2.0 ml of a 48-h culture grown on phosphate-buffered medium with 0.3% CH<sub>3</sub>OH to an OD at 660 nm of 1.1. Liquid was removed for time zero CH<sub>3</sub>OH and radioactivity measurements. Gas samples were then removed from each bottle for H<sub>2</sub> measurements. (An equivalent volume of H<sub>2</sub>-CO<sub>2</sub> was injected and mixed before removing gas and liquid samples.) The <sup>14</sup>CO<sub>2</sub> culture was pressurzed to 202.6 kPa of H<sub>2</sub>-CO<sub>2</sub>. The inoculated cultures were incubated for 48 to 72 h. The total gas volumes and the concentrations of CH<sub>4</sub> and residual H<sub>2</sub> were measured.

After incubation with <sup>14</sup>CH<sub>3</sub>OH, CO<sub>2</sub> and CH<sub>4</sub> were separated essentially by the method of Kunz (3). Briefly, 1 ml of the radioactive culture gas was mixed with cold CO<sub>2</sub> and CH<sub>4</sub> carriers and passed through a liquid N<sub>2</sub> cold trap to retain the CO<sub>2</sub>. The CH<sub>4</sub> was passed through a gas chromatograph equipped with a molecular sieve-charcoal column and thermal conductivity detector. The CO<sub>2</sub> was passed through a separate chromatograph containing a silic gel column (C. Kunz. personal communication). The radioactivity of each separated gas was determined by internal gasproportional spectrometry as described by Paperiello (7).

Five milliliters of gas was removed from each bottle and injected into serum bottles of known volume containing 4.0 ml of phenethylamine and an  $N_2$  atmosphere. The amount of unabsorbed gas in the headspace, measured by displacement of the barrel of a hypodermic syringe, was used to calculate the amount of  $CO_2$  absorbed. For the  $^{14}CO_2$  experiment, the radioactivity of the unabsorbed gas was determined in a toluene-based scintillation fluid (11). The bottles were flushed with  $N_2$  for 2.0 min, and the radioactivity absorbed into the phenethylamine was determined.

Incorporation of <sup>14</sup>C into cell carbon was measured by washing duplicate 6-ml portions of cell culture three times with 6 ml of distilled water. The washed cells were resuspended in 6 ml of water, and 5 ml of each was used for dry weight determinations.

The radioactivity of the resuspended cells, of  $\mathrm{CO}_2$  in phenethylamine, and of the liquid culture medium was determined in Aquasol (New England Nuclear Corp., Boston, Mass.) in a scintillation counter. To determine the radioactivity of the medium after incubation, the cells were first removed by centrifugation. Radioactiv-

ity of unknowns was corrected for quenching by comparing the automatic external standard ratios to those obtained with <sup>14</sup>C-quenched standards.

Other analytical methods. H<sub>2</sub> and CH<sub>4</sub> were measured by gas chromatography as previously described (2). CH<sub>3</sub>OH was determined with a gas chromatograph with a flame ionization detector fitted with a nickel column (6 ft by ½ in. [ca. 182.88 by 0.318 cm]) packed with Chromosorb 101 (80/100 mesh; Supelco). under the following conditions: carrier gas, N<sub>2</sub> (30 ml/min): detector temperature, 155°C; column temperature. 117°C. Unknowns were quantified by comparing their retention times and peak heights with those of standards.

Chemicals. Radioactive NaHCO<sub>3</sub> and CH<sub>3</sub>OH were obtained from New England Nuclear Corp. All gases were at least 99.999% pure. All other chemicals were of reagent grade quality or better.

# RESULTS

The isolate grew and produced  $CH_4$  from  $H_2$  and  $CH_3OH$ , but no growth occurred when  $H_2$  was replaced by  $N_2$  or when  $CH_3OH$  was replaced by  $CO_2$  (Table 1). The small amount of growth with  $H_2$ - $CO_2$  in the absence of methanol was probably due to the presence of small amounts of methanol in the inoculum. (The ODs of the  $N_2$ - $CO_2$  cultures were not significantly different from the ODs of the inoculum.)  $CH_4$  was measured after 9 days of incubation; in cultures grown with  $H_2$ - $CO_2$ , 30 and 3.6  $\mu$ mol/ml were found in the presence and absence of  $CH_3OH$ , respectively. No growth or  $CH_4$  forma-

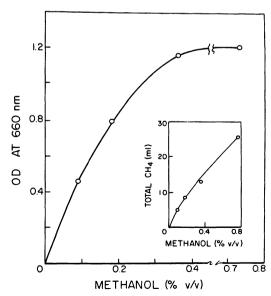


FIG. 2. Growth and  $CH_4$  production with increasing amounts of  $CH_3OH$ . The gas phase was 202.6 kPa of  $H_2$ - $CO_2$ . The highest OD is plotted and was attained in 2 or 3 days. All OD values were corrected for the highest value (0.20 at 48 h) of the zero methanol control.  $CH_4$  was determined after 3 days.

<sup>\*</sup> Final concentration, 0.45% (vol/vol).

tion occurred in the same medium with acetate, methylamine, dimethylamine, trimethylamine, formate, or ethanol with 101.3 kPa of  $H_2$ -CO<sub>2</sub> or  $N_2$ -CO<sub>2</sub> in the gas phase. No growth or  $CH_4$  formation occurred with formate without or with  $CH_3$ OH with  $N_2$ -CO<sub>2</sub>. The isolate did not grow in a complex medium which contained glucose, yeast extract, and Trypticase with 101.3 kPa of 100%  $CO_2$ .

Total growth and the amount of  $CH_4$  formed increased as  $CH_3OH$  increased to a final concentration of approximately 0.4% with  $H_2$  in excess (Fig. 2), or as  $H_2$  increased to approximately 100 kPa with  $CH_3OH$  in excess (Fig. 3). Production of  $CH_4$  continued with excess  $H_2$  or  $CH_3OH$  after growth reached the stationary phase.

The stoichiometry of methanogenesis was determined by measuring the amounts of H<sub>2</sub> and CH<sub>3</sub>OH used and the amount of CH<sub>4</sub> produced. Radioactive <sup>14</sup>CH<sub>3</sub>OH was used to determine the incorporation of the isotope into CH<sub>4</sub> and CO<sub>2</sub>. The amounts of H<sub>2</sub> and CH<sub>3</sub>OH used and CH<sub>4</sub> produced were close to the expected values for a reaction that produces 1 mol of CH<sub>4</sub> from 1 mol each of CH<sub>3</sub>OH and H<sub>2</sub> (Table 2). The specific activity of the CH<sub>4</sub> was 711 dpm/µmol, and that of CH<sub>3</sub>OH was 670 dpm/µmol; no radioactive CO<sub>2</sub> was produced. The fact that CO<sub>2</sub> was not labeled ruled out the possibility of the formation of CH<sub>4</sub> according to equation (1). When the organism was grown with <sup>14</sup>CO<sub>2</sub> with a final specific activity of 1,700 dpm/µmol and

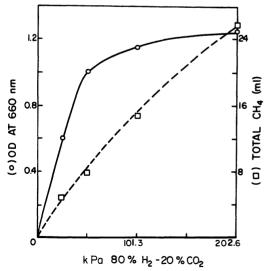


FIG. 3. Growth and  $CH_4$  production with increasing amounts of  $H_2$ - $CO_2$ . Methanol was 0.76%, and  $N_2$ - $CO_2$  was added where necessary to have all gas phases at 202.6 kPa. The highest OD is plotted and was attained at 6 to 7 days of an 8-day incubation period.  $CH_4$  was determined after 8 days.

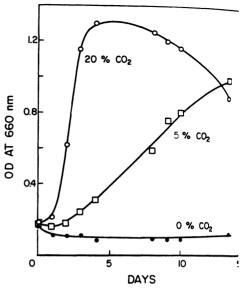


FIG. 4. Requirement for CO<sub>2</sub> for growth. H<sub>1</sub> w. added to bring all gas phases to 101.3 kPa.

unlabeled CH<sub>3</sub>OH, no radioactive CH<sub>4</sub> was formed. The amounts of H<sub>2</sub> and CH<sub>3</sub>OH use. and CH<sub>4</sub> formed were essentially the same for the <sup>14</sup>CO<sub>2</sub> and <sup>14</sup>CH<sub>3</sub>OH experiments. There was significant incorporation of the carbon of CO<sub>2</sub> but not that of CH<sub>3</sub>OH into cells. Based of the specific activity of the CO<sub>2</sub>, the amount of radioactivity incorporated, and an assumption that 50% of the dry weight of cells is carbon. w. estimated that 50% of cell carbon was denvefrom CO<sub>2</sub>. Separate experiments showed the CO<sub>2</sub> was required for growth of the organic (Fig. 4). The initial pH with 20% CO<sub>2</sub> was ca. 115 U lower than with 5% or 0% CO<sub>2</sub> (pH 7.1). The decrease in OD of the 20% CO2 culture appeare. to be due to lysis of the cells.

# DISCUSSION

The organism examined in this study appendix of the control of the

The thermodynamics of the stepwise, two electron reductions of CO<sub>2</sub> to CH<sub>4</sub> indicates the reduction of an intermediate at the same oxidation state as CH<sub>3</sub>OH provides almost all of the energy that can be derived from the reduction of CO<sub>2</sub> to CH<sub>4</sub> with H<sub>2</sub> (9). Because of its restricted

ABLE 2. Stoichiometry of production of CH<sub>4</sub> from CH<sub>3</sub>OH and H<sub>2</sub>

	•	-	
(ompound	Total µmol		
		Expected"	
	Found	Equation (1)	Equation (2)
H <sub>i</sub> OH used	2,720	2,720	2,720
ri used	2,538	0	2.720
H, produced	2,532	2.040	2,720

<sup>-</sup> Based on amount of CH3OH used.

mergy metabolism, the isolate should prove seful for studies of the production of energy by the reduction of a methyl group to CH<sub>4</sub>. The regainsm should also be useful for studies of the procursors because the flow of carbon to H<sub>4</sub> appears to be independent of the flow from O<sub>2</sub> to cell carbon.

In the intestinal habitat, the isolate probably ses for growth the CH3OH produced by other erganisms that degrade pectin (8). H<sub>2</sub> is a normal roduct or intermediate of fermentation by the -uman large intestine microbial community (10). the isolate was obtained from an individual who normally harbors very high concentrations of Methanobrevibacter smithii as the predominant methanogen (5). Although bacteria closely reembling the isolate can be identified by fluorestence microscopy in the most fecal specimens rom the same individual, they are far outnumred by M. smithii and are outgrown by M. mathii on a medium we use to enumerate precommant methanogens in the human large intestine (5). Suitable selective enumeration and isoation techniques will have to be developed to valuate the prevalence and significance of oranisms identical or similar to the isolate in the human large intestine and other methanogenic ecosystems.

#### ACKNOWLEDGMENTS

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